

VIDEO PEMBELAJARAN

MATERI: MAXWELL-BOLTZMANN DISTRIBUTION

PART 2

Dapat diakses di: <http://youtu.be/YAWq87HI9qg> atau
<http://jeikhsan.wordpress.com/kuliah/>

The screenshot shows a YouTube video player for the video 'Jaslin Ikhsan: Maxwell-Boltzmann Distribution Part 2'. The video content is a lecture slide with the following text and formulas:

Evaluate the v_{rms} by use of M-B Distribution!

- The root-mean square speed :
$$v_{rms} = \sqrt{\frac{3KT}{m}}$$
- The average speed:
$$v_{avg} = \sqrt{\frac{8KT}{\pi m}}$$
- The most possible speed:
$$v_{mp} = \sqrt{\frac{2KT}{m}}$$

Remember:

- *You can use K, instead of k, and R, k_B.
- *When you use R, you have to use M (mass of mole) instead of m, and M (m_{ole})_g.

The video player interface shows 255 views, 0 likes, and 0 dislikes. The video was uploaded by 'jaslinikhsan' on May 25, 2011. The description reads: 'To study the Kinetic Theory of Gas. Using the equation of Maxwell-Boltzmann'. The video player is embedded in a Mozilla Firefox browser window. The browser's address bar shows the URL: <http://www.youtube.com/watch?v=YAWq87HI9qg>. The browser's tabs include 'jaslin ikhsan - YouTube', 'Bahan Kuliah - Jaslin Ikhsan Homep...', 'Jaslin Ikhsan: Chemical Kinetic Part ...', and 'Jaslin Ikhsan: Maxwell-Boltzmann...'. The browser's search bar contains 'jaslin ikhsan'. The browser's status bar at the bottom shows 'Transferring data from o-o-preferred.pttelkom-bth1.v21.lscache3.c.youtube.com...'. The system tray at the bottom right shows the time as 6:25.

Jaslin Ikhshan: Maxwell-Boltzmann Distribution Part 2 - YouTube - Mozilla Firefox

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$$\frac{dN}{N} = 4\pi v^2 \times \left(\frac{m}{2\pi kT}\right)^{3/2} \times e^{-\frac{1}{2}mv^2/kT} \cdot dv$$

$$\frac{dN}{N} = f(v) = 4\pi v^2 \times \left(\frac{m}{2\pi kT}\right)^{3/2} \times e^{-\frac{1}{2}mv^2/kT}$$

03:37 / 14:11 360p

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Uploaded by jaslinikhshan on May 25, 2011

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